

Insight into the contributions of primary emissions of sulfate,
nitrate, and ammonium from residential solid fuels to ambient PM_{2.5}

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Abstract

Understanding the primary emissions of sulfate (SO_4^{2-}), nitrate (NO_3^-), and ammonium (NH_4^+) (SNA) from solid fuels (coal and biomass) combustion is important to study their roles in haze formation and particle growth. In this study, direct emissions of SNA and other inorganic ions (including Na^+ , K^+ , Mg^{2+} , Ca^{2+} , and Cl^-) in fine particulate matter ($\text{PM}_{2.5}$) from residential coal combustion (RCC) and biomass burning (BB) were quantified through combustion chamber experiments. Emission factors (EFs) of the total quantified ions for the five types of solid fuels are in the range of 178-3,880 mg/kg, accounting for 5.8%-41.1% of the emitted $\text{PM}_{2.5}$ mass. The average proportions of SO_4^{2-} , NO_3^- , and NH_4^+ in $\text{PM}_{2.5}$ emitted from RCC are 3.7%, 0.9%, and 1.0%, respectively, in comparison to 1.3%, 0.8%, and 0.1%, respectively, for BB. Despite the variations of SNA proportions seen among the solid fuel types, SO_4^{2-} is the most dominating inorganic ion, consistent with the emission profiles shown in other literatures. Similar mass fraction and its range of SO_4^{2-} are found between RCC and ambient in the northern cities, implying that the primary emission from RCC is a significant source contributor to atmospheric SO_4^{2-} , particularly in wintertime. According to the EFs and mass fractions of SNA determined for the solid fuels, the contribution of secondary formation to atmospheric SO_4^{2-} should be overestimated in ambient $\text{PM}_{2.5}$ source apportionment.

Keywords: $\text{PM}_{2.5}$; Sulfate; Nitrate; Ammonium; Solid Fuels; Source Identification

1. Introduction

Particulate matter (PM) has been emphasized for its impacts on both human health and climate change (Cao et al., 2012b; Fu et al., 2016; Kelly and Fussell, 2012; Shen et al., 2007; Wang et al., 2017). In 2012, the China Government promulgated the national standard of the annual average concentration of PM_{2.5} (PM with an aerodynamic diameter of equal to or less than 2.5 μm) of 35 μg/m³ (Cao et al., 2012a; Kong et al., 2018). A full understanding of PM_{2.5} compositions and sources is essential for establishing efficient pollution control measures (Kong et al., 2010). Ambient PM_{2.5} typically consists of inorganic ions, organic carbon (OC), element carbon (EC), geological minerals, salts, trace elements, and other substances (Chow et al., 2015). The sum of inorganic ions contributed significantly to PM_{2.5} in northwest China (Niu et al., 2016). Sulfate (SO₄²⁻), nitrate (NO₃⁻), and ammonium (NH₄⁺) (denoted as SNA) are the three most abundant inorganic components (Shen et al., 2008).

Atmospheric SNA could be originated from both primary emissions and secondary formation. Typical primary sources include coal combustion, biomass burning, fugitive dust, industrial, and vehicular emissions (Gao et al., 2011; Shen et al., 2016; Švédová et al., 2020; Zhang et al., 2015; Zhou et al., 2016). The secondary formation of atmospheric SNA is recognized as oxidation and neutralization products from major gaseous precursors, including sulfur dioxide (SO₂), nitrogen oxides (NO_x), and ammonia (NH₃) (Wang et al., 2013; Zhang et al., 2015). The concentrations levels of SNA from the secondary formation are considered one of the main driving forces in PM_{2.5} pollution (Huang et al., 2014). However, research on sulfate formation mechanisms by atmospheric chemistry models has improved our understanding of secondary sulfate, but gaps still existed between observation and model prediction in both haze and non-haze days (Wang et al., 2014). The gaps could attribute to missing sources, such as primary emissions. For example, Shen et al (2008) found that even with a much lower sulfur oxidation ratio (SOR) and similar nitrogen oxidation ratio (NOR) observed in winter compared to summer, much lower ratios of NO₃⁻/SO₄²⁻ still

occurred in winter in Xi'an, northwest China, implying that the large quantities of atmospheric SO_4^{2-} directly emitted from coal combustions for heating. Dai et al. (2019) reported that primary emissions from residential coal combustion in rural Xi'an accounted for as high as 49.2% of total atmospheric SO_4^{2-} in winter. Thus, using SO_4^{2-} as a secondary formation marker in source apportionment studies could have large uncertainties, especially in the regions where solid fuels such as coal were extensively used.

To fill the knowledge gap discussed above, combustion experiments were carried out using four types of coal and one type of biomass (maize straw) that were commonly used in northwest China. Emission factors (EFs) of SNA and their proportions in $\text{PM}_{2.5}$ emitted from residential coal combustion (RCC) and biomass burning (BB) were determined. Furthermore, the results from the present study were compared with literature-reported data on the SNA proportions in $\text{PM}_{2.5}$ emitted from a variety of fuels under various combustion conditions. Contributions from primary SNA emissions to ambient $\text{PM}_{2.5}$ were finally elaborated.

2. Methods

2.1 Fuel selection and combustion tests

Four types of coals, including anthracite chunk (A-chunk), anthracite briquette (A-briquette), bituminous chunk (B-chunk), and bituminous briquette (B-briquette), and one biomass fuel of maize straw were selected as the combustion samples considering their wide applications in northwest China. The briquettes used in this study were produced following the procedures described in our previous study (Zhang et al., 2020). During the briquetting process, no additive was added to avoid the alternation of coal properties and particle emission characteristics.

The experiments were all conducted inside a regulated combustion chamber with a volume of 8 m^3 . The chamber and inner surface are made with aluminum alloy and equipped with a dilution sampler which captured emitted particles from the stages of

ignition to extinction. Detailed descriptions of the chamber were shown by Sun et al. (2017) and Zhang et al. (2020). The dilution sampler can collect PM_{2.5} in the diluted plume onto three 47 mm filters in parallel, including two quartz filters (Whatman, Maidstone, UK) and one Teflon filter (Pall Life Sciences, Ann Arbor, MI, USA) (Tian et al., 2015). The gravimetric measurement was conducted using an electronic balance with a sensitivity of $\pm 1 \mu\text{g}$ (ME 5-F, Sartorius, Göttingen, Germany). All of the filters before and after the sample collection were placed in a weighing chamber at a temperature of $22 \pm 2 \text{ }^\circ\text{C}$ and relative humidity (RH) of $35\% \pm 5\%$ for at least 24 h before weighing.

2.2 Water-soluble ions analysis

One-quarter of the quartz filter was cut, placed into a Teflon tube, and extracted with 10 ml of deionized water by mechanical shaking for 30 min, and repeated extraction three times. The combined extractant was then filtered through a $0.45 \mu\text{m}$ pore size membrane before being injected into an ion chromatography (Dionex Cooperation, Sunnyvale, CA, USA). Cations (i.e., Na^+ , NH_4^+ , K^+ , Mg^{2+} , and Ca^{2+}) and anions (i.e., Cl^- , NO_3^- , and SO_4^{2-}) were quantified. Detailed information on the ion analytical method and its QA/QC can be found in Shen et al (2011).

2.3 Emission factors calculation

Mass-based emission factors (EFs) were calculated using the following equation (Zhang et al., 2020):

$$\text{EF} = \frac{m_{\text{particle}} \times \text{DR} \times t_{\text{sample}} \times V_{\text{stk}} \times D}{Q_{\text{filter}} \times m_{\text{fuel}}} \quad (1)$$

where m_{particle} stands for the mass of particles deposited on the filter (mg), DR is the dilution rate during sampling, t_{sample} is the sampling time (from ignition to extinction); V_{stk} is the velocities of the chamber stack (m/s), D stands for the cross-sectional area of the stack (m^2); Q_{filter} is the volume of air passed through the filter (m^3), and m_{fuel} is the mass of fuel consumed in each test (kg).

3. Results and Discussion

3.1 PM_{2.5} and water-soluble ions emitted from RCC and BB

Table 1 lists the EFs of PM_{2.5} and the target inorganic ions for the four types of coal and one type of biomass fuel. The PM_{2.5} EFs for the four coals range from 0.51±0.13 to 3.08±0.19 g/kg, much lower than that of 11.2±3.67 g/kg for maize straw (BB). The EFs of the total quantified ions range from 178-3,886 mg/kg for both solid fuels, contributing 5.8%-41.1% of the PM_{2.5} mass. For the anthracite coal, the EFs of SO₄²⁻, NO₃⁻, and NH₄⁺ for its chuck/briquette are 29.9±7.52/76.2±71.8, 8.86±1.46/8.51±0.00, and 21.0±24.8/18.4±24.1 mg/kg, respectively, while for the bituminous coal, showing 27.8±3.70/36.9±19.2, 9.44±1.16/8.44±4.27, 0.04±0.06/6.47±4.00 mg/kg, respectively. Even though EFs of SO₄²⁻ and NO₃⁻ between the chuck and briquette for anthracite and bituminous coals are close, large discrepancies are seen for other inorganic ions. For instance, the EFs of NH₄⁺ for briquette bituminous are three orders of magnitude higher than raw bituminous, demonstrating the compositions of inorganic ions could be influenced by the briquetting process. In comparison to the coals, much higher EFs of SNA are seen for maize straw, which are 108±95.0, 68.0±30.3, and 10.9±17.9 mg/kg of SO₄²⁻, NO₃⁻, and NH₄, respectively. The results demonstrate that variations in fuel types could significantly alter the EFs for SNA, and further change the contribution rate of initial emission from RCC and BB to the atmospheric occurrence, particularly during the wintertime for heating using solid fuel. Rather than SNA, the EFs of K⁺ for maize straw is 2,150±1,586 mg/kg, much higher than that of other ions. It should be also noted that the EFs of K⁺ for coals are approximately three orders of magnitude lower than for maize straw, which accounted for a negligible fraction of the total quantified inorganic ions. The finding is consistent with the fact that K/K⁺ is a remarkable tracer for BB in source apportionment (Andreae and Merlet, 2001; Chen et al., 2017; Sun et al., 2019).

Insert Table1

3.2 Proportions of SNA in PM_{2.5} emitted from solid fuels in this study and comparison with other sources in the literature

Table 2 lists the mass proportions of SNA in PM_{2.5} emitted from the solid fuels of RCC and BB in this study. The average proportions of SO₄²⁻, NO₃⁻, and NH₄⁺ in PM_{2.5} for RCC are 4.1±3.4%, 1.1±1.0%, and 1.4±2.1%, respectively, in comparison to lower values of 1.3±1.4%, 0.8±0.5%, 0.1±0.1% for BB. The results probably represent the primary SNA emission is more predominant in RCC than BB. Furthermore, SO₄²⁻ for RCC shows the highest ion proportion in PM_{2.5} (Dai et al., 2019; Shen et al., 2008), which is consistent with its high abundance in northwest China, especially during the heating season (Shen et al., 2008). It should be noted that significantly lower proportions of NH₄⁺ and NO₃⁻ ($p<0.05$) are found for RCC, demonstrating that it is not a significant source for these two inorganic ions.

Insert Table2

Figure 1 further illustrates the proportions of SNA emitted from RCC and BB among the studies (as the data shown in Table 2) to present the range of SNA fractions in PM_{2.5} for RCC and BB. The mass proportions of SNA for both RCC and BB follow the order of SO₄²⁻ > NH₄⁺ > NO₃⁻. Among those studies, the average mass proportions of SO₄²⁻, NO₃⁻, and NH₄⁺ in PM_{2.5} for RCC are 10.1%, 0.7%, and 2.0%, respectively, whereas lower values of 1.1%, 0.3%, and 0.3% are seen for BB. In terms of range, much smaller variations are seen for NO₃⁻ (0.05%-2.2%) than SO₄²⁻ (1.0%-30.0%) and NH₄⁺ (0.003%-9.9%). These fractions could be largely inferred by the physical states (e.g., chunk or briquette) and coal sources (anthracite or bituminous) on the emission of SNA. For BB, the proportions of SO₄²⁻, NO₃⁻, and NH₄⁺ in PM_{2.5} are in ranges of 0.1%-6.9%, 0.03%-0.8%, and 0.03%-1.7%, respectively. The SNA proportions in BB PM_{2.5} all show much narrower variations compared to that of RCC, which reveals that the SNA emissions are relatively much more stable from BB than from RCC. And the SNA

abundances in BB are relatively lower compared to that of RCC.

Insert Figure1

Table 2 also summarizes EFs and proportions of SNA in $PM_{2.5}$ emitted from a variety of solid fuels and materials (e.g., peats, electric waste, household garbage, barbeque, road dust, and construction) using different combustion devices and under various combustion conditions. The EFs and proportions of NH_4^+ and NO_3^- in $PM_{2.5}$ emitted from open burnings of electric waste (E-waste) and household garbage burning, barbeque, and dust are close to the values for RCC. However, the EFs and proportion of SO_4^{2-} are much higher than those of other sources. Besides, the SNA proportions for BB are similar to those of dust but far lower than those of open-burning activities.

Results before and after aging from peat burning reported by Chow et al. (2019) was also compared. Peat is an important and common solid fuel. Its EFs and proportions of SNA for five different peats collected worldwide are all lower than those of RCC and BB in this study. However, after simulations of two-equivalent days of atmospheric aging with an oxidation flow reactor (OFR), the proportions of NH_4^+ and NO_3^- in $PM_{2.5}$ emitted from peats are higher than those from BB. And after seven-equivalent days of aging, the two ions are even higher than those from RCC. Eventually, the secondary formations occur with their corresponding gaseous precursors (i.e., NH_3 and NO_x) during the aging processes. Nevertheless, the proportions of SO_4^{2-} in $PM_{2.5}$ emitted from peats are still lower than those of RCC in this study even after seven -equivalent days of stimulated aging. This concludes that the secondary formation of SNA might not be always more important than primary emission for a few solid fuels, particularly of SO_4^{2-} in coal in this study. The limitation of its precursor (e.g., SO_2) would be a dominant factor.

3.3 Comparisons of SNA proportions in ambient $PM_{2.5}$

Table 3 summarizes the SNA proportions in ambient $PM_{2.5}$ in the Chinese cities,

and Figure 2 compares the statistical data (i.e., average value). The range (average) proportions of SO_4^{2-} , NO_3^- , and NH_4^+ in ambient $\text{PM}_{2.5}$ are 6.1%-36.0% (19.1%), 4.8%-36.9% (15.1%), and 3.2%-19.7% (9.9%), respectively. The general order of $\text{SO}_4^{2-} > \text{NO}_3^- > \text{NH}_4^+$ in ambient $\text{PM}_{2.5}$ is different from those of RCC and BB (i.e., $\text{SO}_4^{2-} > \text{NH}_4^+ > \text{NO}_3^-$). For NH_4^+ , the lowest proportion in ambient $\text{PM}_{2.5}$ is even higher than the median for RCC or BB in this study (Table 2), implying that there are other critical sources for atmospheric NH_4^+ . For NO_3^- , the lowest proportion in ambient $\text{PM}_{2.5}$ is also much higher than that for RCC and BB, suggesting that primary emissions from RCC and BB are not the major contributor of NO_3^- in most Chinese cities.

The proportion of SO_4^{2-} for BB is relatively low, but a higher value is seen in ambient $\text{PM}_{2.5}$ (19.1% on average), implying that BB is not a significant source. On the contrary, it is noted that the ranges of SO_4^{2-} proportion are similar between RCC (1.0%-30.0%) and ambient $\text{PM}_{2.5}$ (6.1%-36.0%), but the medium is much lower for RCC (5.3%) than ambient $\text{PM}_{2.5}$ (17.5%). This could be explained by various pollution sources contributed to ambient SO_4^{2-} . However, the primary emission of RCC could be a major source of SO_4^{2-} in the northern cities in China, such as Lanzhou, Xi'an, Beijing, and Tianjin, which show an average mass proportion of 13.3%, consistent with the average proportion of RCC. A series of research conducted by Dai et al. (2018; 2019) proved that atmospheric sulfate in winter Xi'an was not only attributed to secondary formation. Moreover, the $\text{PM}_{2.5}$ -bound sulfate is 49.2% from RCC primary emissions and 50.8% from SO_2 -derived secondary sources and other primary sources in winter Xi'an. For the southern cities in China, the impacts of RCC should be much lower due to less solid fuel combustion in wintertime, but a higher proportion of SO_4^{2-} (23.0% on average), implying that more potential secondary formation occurred in this region.

Insert Figure2

Insert Table3

According to Table S1, the contribution rate (less than 1%) from BB primary emissions to ambient SNA are all negligible in either winter or summer, south or north.

For RCC primary emissions, no significant contribution rates were found for ambient nitrate as well (0.1%-3.0%). But the contribution rates to ambient ammonium (3.4%-9.3%) in northern cities in winter cannot be ignored. And RCC primary emissions significantly contributed to ambient sulfate (10.5%-26.6%) in northern cities in winter. While in southern cities, or northern cities in summer, residential coal was not widely used. Therefore, no similar results can be found.

Uncertainties do exist in our estimation and mainly originated from two sources, and uncertainties for estimated results in Table S1 should be the combination of the two sources. Source one is the uncertainty in sulfate EFs obtained from literatures that may not be harmonized when measuring the composition. Ideally, such kind of uncertainties that related to measuring methods and quality control measures should be well expressed in variations of EFs. Thus, average sulfate EFs used for estimation were 10.1 for RCC and 1.1 for BB (Fig 1), with type A uncertainties (expressed as percentages) of 24.4%, 26.4%, respectively. Source two is from meteorological conditions that affect physical process like advection and diffusion of primary air pollutants but also formation of secondary air pollutants. Due to lack of uncertainty numbers in the cited papers regarding source two, detailed estimation cannot be done here. Thus, the overall uncertainties for sulfate concentrations in Table S1 are 24.4% for RCC and 26.4% for BB, respectively. However, uncertainties from source two could also alter the estimated sulfate concentrations significantly and should be well quantified in future source apportionment studies.

4. Conclusions

EFs of primary SNA in $PM_{2.5}$ were determined for the common solid fuels used for heating in northwest China. Compared with the results in the literatures, the proportions of SNA for RCC greatly vary with different physical states and types of the coals, in addition to the stoves and combustion conditions. The emissions of SNA for BB are comparatively stable but lower than RCC. The largest variation in EFs and mass fraction is observed on the SO_4^{2-} from RCC, consistent with the contributions in

ambient PM_{2.5}. Primary emission from RCC could be a significant contributor to atmospheric SO₄²⁻ in northwest China, which are more impacted by RCC and high precursor gas emission from coals during the heating season with low SOR. Even though SO₄²⁻ is a dominating water-soluble ion in PM_{2.5} in China in the past decades, previous studies might overestimate its contribution to secondary formation in source apportionment. The results of the present study should assist in adjusting the bias and provide more understanding of their formations as well as establishing more accurate and efficient PM_{2.5} control measure in the future. This study highlights that it is importance to control the primary sulfate emissions on air quality management.

Acknowledgment

This research was supported by the Key R&D Project of Shaanxi Province, China (2022ZDLSF06-07) and the Science and Technology plan project of Xi'an city (21NYYF0041).

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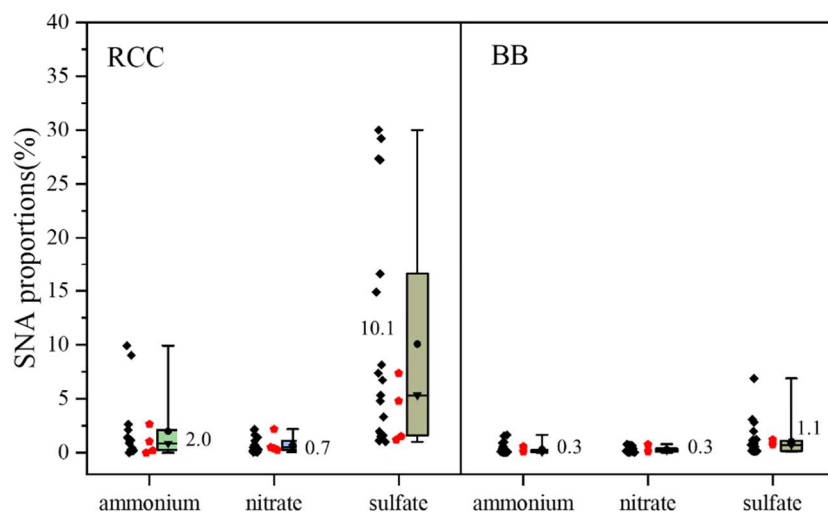
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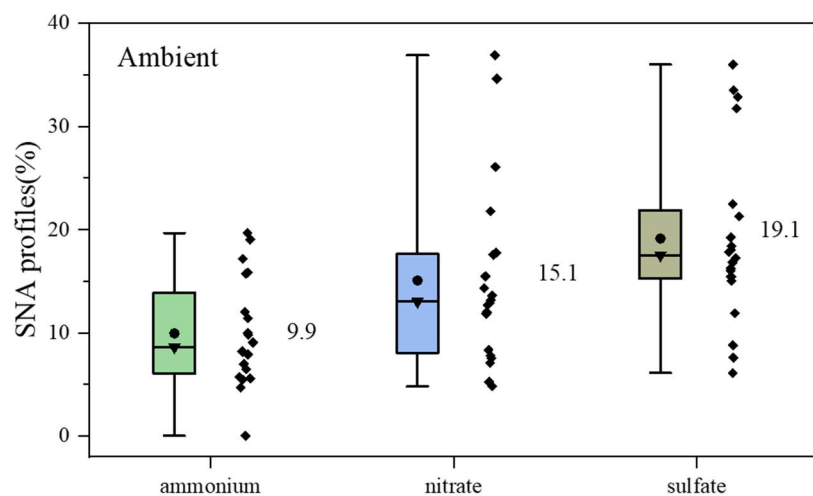
478 **Figure Captions**

479 Figure 1. A summary of SNA proportions in PM_{2.5} emitted from RCC and BB in
480 literatures (black point) and this study (red point). The box plot represents the 25th and
481 75th percentile, and the lower and upper whiskers represent the lowest and highest
482 values. Solid circles and solid triangles represent the average value and median value,
483 respectively, for each proportion.

484 Figure 2. SNA proportions obtained from ambient studies. Box plot represent the 25th
485 and 75th percentile and lower and upper whiskers represent lowest and highest values.
486 Solid circle and solid triangle represent the average value and median value for each
487 kind of proportions.



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 490 Figure 1. A summary of SNA proportions in PM_{2.5} emitted from RCC and BB in
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 492 75th percentile, and the lower and upper whiskers represent the lowest and highest
 493 values. Solid circles and solid triangles represent the average value and median value,
 494 respectively, for each proportion.
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 498 Figure 2. A summary of SNA proportions in ambient PM_{2.5} from literature. The box
 499 plot represents the 25th and 75th percentile, and the lower and upper whiskers
 500 represent the lowest and highest values. Solid circles and solid triangles represent the
 501 average value and median value, respectively, for each proportion.
 502

503 Table 1. EFs of PM_{2.5}, OC, EC, water-soluble ions, and elements from residential coal
504 combustion and biomass burning obtained in this study; and the results of fuel
505 proximate analysis

Component	Solid Fuels*				
	A-chunk	A-briquette	B-chunk	B-briquette	Maize straw
Unit: g/kg					
PM _{2.5}	0.51±0.13	1.40±0.57	2.35±0.87	3.08±0.19	11.2±3.67
Unit: mg/kg					
Na ⁺	61.2±27.4	52.9±21.6	100.8±1.59	54.6±6.98	525±209
NH ₄ ⁺	21.0±24.8	18.4±24.1	0.04±0.06	6.47±4.00	10.9±17.9
K ⁺	2.77±1.86	1.96±0.06	2.82±0.11	3.48±2.18	2150±1586
Mg ²⁺	6.57±4.15	6.08±5.03	13.8±0.05	4.53±0.91	95.8±62.9
Ca ²⁺	41.8±10.1	57.6±34.2	116.6±20.2	51.8±15.7	638±244
Cl ⁻	37.5±35.2	64.1±61.7	17.9±0.75	11.9±3.81	290±228
NO ₃ ⁻	8.86±1.46	8.51±0.00	9.44±1.16	8.44±4.27	68.0±30.3
SO ₄ ²⁻	29.9±7.52	76.2±71.8	27.8±3.70	36.9±19.2	108±95.0
Σions	210±112	286±219	289±27.7	178±57.1	3886±2474
Σions/PM _{2.5} (%)	41.1	20.4	12.3	5.8	34.7
Moisture (%)	0.4	1.2	9.2	6.5	6.1
Ash (%)	8.6	7.9	3.5	4.8	4.7
Volatile matter (%)	7.3	10.1	25.7	31.2	76.0

506 * A-chunk and A-briquette represent anthracite chunk coal and anthracite briquette coal, respectively;

507 B-chunk and B-briquette represent bituminous chunk coal and bituminous briquette coal, respectively

508 Table 2. EFs of PM_{2.5}, SO₄²⁻, NO₃⁻, NH₄⁺ and the ion proportions to PM_{2.5} for different
509 solid fuels obtained in this and other studies

	EFs (g/kg)	EFs (mg/kg) /proportion in PM _{2.5} (%)			
*Fuel/Combustion style	PM _{2.5}	NH ₄ ⁺	NO ₃ ⁻	SO ₄ ²⁻	Reference
Coal					
A-chunk/TS	0.51±0.13	21.0±24.8 / 2.7±3.3	8.86±1.46 / 2.2±0.7	29.9±7.52 / 7.4±2.8	This study
A-briquette/TS	1.40±0.6	18.4±24.1 / 1.1±1.3	8.51±1.13 / 0.4±0.6	76.2±71.8 / 4.8±3.2	
B-chunk/TS	2.35±0.87	0.04±0.06 / 0.003±0.005	9.44±1.16 / 0.5±0.3	27.8±3.7 / 1.5±0.7	
B-briquette/TS	3.08±0.19	6.47±4.00 / 0.2±0.1	8.44±4.27 / 0.3±0.2	36.9±19.2 / 1.2±0.7	
Coal/TS	ND	ND / 1.4±1.3	ND / 0.3±0.3	ND / 3.3±2.8	(Watson et al., 2001)
A-briquette/TS	ND	ND / 9.0±3.8	ND / 1.7±1.2	ND / 6.7±3.7	(Zhang et al., 2012)
Semi coke/TS	0.75±0.19	ND / ND	3.89±0.77 / 0.5	204±83 / 27.2	(Tian et al., 2018)
B-chunk/TS	11.1±8.13	ND / ND	84.3±22.7 / 0.8	204±74 / 1.8	
A-chunk/TS	0.45±0.09	ND / ND	9.55±1.09 / 2.1	123±54 / 27.3	
A-briquette/TS	1.21±0.44	ND / ND	17.4±4.77 / 1.4	363±162 / 30.0	
A-briquette/TS	2.95	ND / 2.1±0.0	ND / 0.39±0.02	ND / 16.6±5.7	(Dai et al., 2019)
B-chunk/TS	8.58	ND / 9.9±0.5	ND / 1.12±0.2	ND / 29.2±6.6	
^a Honeycomb-1/TS	4.88	11.59 / 0.2	4.96 / 0.1	53.45 / 1.1	(Yan et al., 2020)
^a Honeycomb-2/TS	12.5	53.9 / 0.4	6.06 / 0.1	129 / 1.0	
^a Honeycomb-3/TS	2.84	25.0 / 0.9	1.49 / 0.1	56.6 / 2.0	
^a A-chunk/TS	0.39	4.52 / 1.2	2.25 / 0.6	20.8 / 5.3	
^a B-chunk/TS	5.12	15.9 / 0.3	9.09 / 0.2	82.8 / 1.6	
^a Honeycomb/TS	3.32±2.43	8.9±3.8 / 0.3	10.9±7.2 / 0.3	494±269 / 14.9	(Yan et al., 2017)
^a A-chunk/TS	1.29±0.64	2.3±4.0 / 0.2	8.3±5.1 / 0.6	105±5.1 / 8.1	
Biomass					
Maize straw/TS	11.2±3.67	10.9±17.9 / 0.1±0.1	68.0±30.2 / 0.8±0.5	108±95.0 / 1.3±1.4	This study
^a Maize straw/HK	46.1±1.4	201 / 0.4	324 / 0.7	536 / 1.2	(Sun et al., 2017)
Wheat straw/IS	9.93±0.34	60.2±11.9 / 0.6±0.1	17.9±2.98 / 0.2±0.0	72.0±12.0 / 0.7±0.1	(Sun et al., 2019b)
Wood branch/IS	1.59±0.20	2.05±0.78 / 0.1±0.1	1.41±0.56 / 0.1±0.0	14.3±2.96 / 0.9±0.2	
Wood/TS	ND	ND / 0.1±0.0	ND / 0.1±0.0	ND / 0.9±0.4	(Watson et al., 2001)
Red Maple/TS	0.88±0.16	ND / 0.2±0.0	ND / 0.7±0.1	ND / 0.6±0.1	(Fine et al., 2004)
White Oak/TS	3.4±0.5	ND / 0.1±0.0	ND / 0.4±0.0	ND / 1.0±0.0	
Loblolly Pine/TS	2.0±0.3	ND / 0.3±0.0	ND / 0.2±0.1	ND / 0.2±0.0	
^a Q. pyrenaica/TS	13.4±3.8	14.4±1.5 / 0.1	17.1±2.2 / 0.1	16.8±7.7 / 0.1	(Calvo et al., 2015)
^a P. nigra/TS	4.4±1.4	2.7±1.5 / 0.1	12.1±0.55 / 0.3	47.8±9.9 / 1.1	
^a F. sylvatica/TS	2.82±0.70	2.78±0.44 / 0.1	7.1±1.9 / 0.3	8.2±2.3 / 0.3	
^a Q. pyrenaica/open	12.5±4.1	9.8±4.9 / 0.1	13.5±1.3 / 0.1	16.9±4.1 / 0.1	
^a P. nigra/open	14.0±5.1	4.8±1.3 / 0.0	7.6±1.3 / 0.1	23.2±4.0 / 0.2	

^a F. sylvatica/open	5.8±1.3	2.21±0.40 / 0.0	10.4±0.46 / 0.1	14.8±3.0 / 0.3	
Forest fire/open	ND	ND / 0.1±0.1	ND / 0.0±0.0	ND / 0.1±0.1	(Watson et al., 2001)
Yellow Poplar/open	6.8±0.8	ND / 0.0±0.0	ND / 0.3±0.0	ND / 0.4±0.0	
White Ash/open	3.3±0.3	ND / 0.1±0.0	ND / 0.7±0.0	ND / 0.8±0.1	
Sweetgum/open	3.5±0.4	ND / 0.1±0.0	ND / 0.6±0.0	ND / 0.5±0.0	(Fine et al., 2002)
Mockernut Hickory/open	6.8±0.9	ND / 0.1±0.0	ND / 0.3±0.0	ND / 0.2±0.0	
Loblolly Pine/open	3.7±0.4	ND / 0.1±0.0	ND / 0.3±0.0	ND / 0.2±0.0	
Slash Pine/open	1.6±0.3	ND / 0.2±0.0	ND / 0.4±0.1	ND / 1.1±0.1	
^a Wheat straw/open	11.4±4.9	180±90 / 1.6	22±11 / 0.2	86±79 / 0.8	
^a Rice Straw/open	8.5±6.7	140±100 / 1.7	29±15 / 0.3	240±160 / 2.8	(Ni et al., 2017)
^a Corn straw/open	12.0±5.4	120±120 / 1.0	21±12 / 0.2	240±70 / 2.0	
^a Sugarcane straw/open	ND	ND / 0.0	ND / 0.1	ND / 6.9	(Chuang et al., 2019)
^a Sorghum straw/open	ND	ND / 0.9	ND / 0.7	ND / 3.1	
Other sources					
^b Odintsovo peat (P)/open	ND	ND / 0.1	ND / 0.2	ND / 0.2	
Odintsovo peat (A-2)/open	ND	ND / 1.1±0.9	ND / 0.7±0.1	ND / 0.7±0.5	
Odintsovo peat (A-7)/open	ND	ND / 3.4±1.4	ND / 2.0±0.7	ND / 0.8±0.2	(Chow et al., 2019)
^b Malaysia peat (P)/open	ND	ND / 0.0	ND / 0.1	ND / 0.2	
Malaysia peat (A-2)/open	ND	ND / 0.8±0.1	ND / 0.9±0.2	ND / 0.6±0.2	
Malaysia peat (A-7)/open	ND	ND / 4.7±0.8	ND / 4.7±1.3	ND / 2.0±0.1	
Electric waste/open	25.4±17.0	ND / 0.6±1.0	ND / 0.1±0.1	ND / 2.5±1.6	
Household garbage/open	18.4±9.78	ND / 2.3±0.1	ND / 0.7±0.1	ND / 1.4±0.3	(Wang et al., 2020)
Barbeque/open	16.8±1.87	ND / ND	ND / 1.0±0.9	ND / 3.4±2.8	
^a Road dust	ND	ND / ND	ND / 0.4	ND / 1.7	(Shen et al., 2016)
^a Construction dust	ND	ND / ND	ND / 0.3	ND / 1.5	
Road dust	ND	ND / ND	ND / 0.2	ND / 1.3	(Sun et al., 2019)
Construction dust	ND	ND / ND	ND / 0.2	ND / 1.2	
Road dust-Downtown	ND	ND / 0.02	ND / 0.6	ND / 1.2	(Sun et al., 2021)
Road dust-Non downtown	ND	ND / 0.03	ND / 0.2	ND / 0.8	

510 * EFs are expressed in three significant digits; proportions are expressed in one decimal place, and “ND”
511 denotes no data available in the cited literature; “TS” denotes “Traditional Stoves”; “HK denotes “Heated
512 Kang”, “IS” denotes “Improved Stoves”.

513 ^a Proportions were calculated using the EFs for the ions and PM_{2.5} obtained in the corresponding literature.

514 ^b The average value accounted for all primary proportions shown in the literature.

Table 3. Ambient mass concentrations of $\text{PM}_{2.5}$, SO_4^{2-} , NO_3^- , and NH_4^+ in selected Chinese cities

City	Season	Conc ($\mu\text{g}/\text{m}^3$)				References
		$\text{PM}_{2.5}$	NH_4^+	NO_3^-	SO_4^{2-}	
Xi'an (urban)	Annual	167	9.13	19.8	25.8	(Wang et al., 2015)
Xi'an (rural)	Annual	90.2	6.30	14.0	17.4	(Wang et al., 2015)
Xi'an	Winter	345	16.2	24.5	26.2	(Huang et al., 2014)
Beijing	Winter	159	15.6	19.1	25.4	(Huang et al., 2014)
Lanzhou	Winter	182	11.7	14.1	16.0	(Tan et al., 2017)
Shanghai	Winter	91.0	7.46	12.4	10.8	(Huang et al., 2014)
Guangzhou	Winter	69.0	6.90	8.90	12.7	(Huang et al., 2014)
Nanjing	Winter	63.0	12.0	21.8	20.7	(Hua et al., 2015)
Suzhou	Winter	99.0	17.0	25.8	31.4	(Hua et al., 2015)
Tianjin and Wuqing	Annual	149	8.50	19.6	24.2	(Zhou et al., 2016)
Zhejiang and Haining	Annual	110	6.10	13.9	16.5	(Zhou et al., 2016)
Taichung, Taiwan	Annual	21.0	4.13	4.57	7.56	(Young et al., 2016)
Beijing	Summer	138	10.9	11.5	23.2	(Tao et al., 2016)
Beijing	Winter	139	4.5	7.3	8.5	(Tao et al., 2016)
Hangzhou	Annual	80.0	9.10	14.2	13.8	(Xu et al., 2017)
Hefei	Annual	86.3	7.82	15.1	15.56	(Deng et al., 2016)
Chongqing (urban)	Summer	47.9	3.90	2.30	10.2	(Qiao et al., 2019)
Chongqing (urban)	Winter	90.7	10.9	13.0	20.4	(Qiao et al., 2019)
Wuhan	Average of Jun to Aug	25.6	4.02	1.92	8.57	(Xie et al., 2020)
Wuhan	Average of Dec to Feb	66.0	10.4	24.3	11.8	(Xie et al., 2020)

** Concentrations are expressed in three significant digits; proportions are expressed in one decimal place.